

#### DKS 2501:2013

#### **TECHNICAL COMMITTEE REPRESENTATION**

The following organizations were represented on the Technical Committee:

PZ Cussons E.A. Kenya Industrial Research and Development Institute (KIRDI) Government Chemist's Department Orbit Chemicals Odex Chemicals Ltd Reckitt Benckiser Johnson Diversey E & C Africa University of Nairobi Ecolab Limited Rumorth Group MEP Chemical Industries Consumer Information Network Kenya Bureau of Standards — Secretariat

#### REVISION OF KENYA STANDARDS

In order to keep abreast of progress in industry, Kenya Standards, hall be regularly reviewed. Suggestions for improvements to published standards, addressed to the Manacing un cor, Kenya Bureau of Standards, are welcome.

### © Kenya Bureau of Standa ds, 2013

Copyright. Users are reminded that by virtue of section of the Copy on the Act, Cap. 130 of the Laws of Kenya, copyright subsists in all Kenya Standards and except as provided under section 7 of this Act, no Kenya Standard produced by Kenya Bureau of Standards may be reproduced, stored in a retrieval system in any form or transmitted by a virtual without prior permission in writing from the Managing Director.

Permission may be conditional on an appropriate roya.

Care should be taken to ensure that material user is non-the current edition of the standard and that it is updated whenever the standard is amended or revised. The number and date of the standard should therefore be clearly identified.

pav

The use of material in print or in electonic form to be used commercially with or without payment or in commercial contracts is subject to payment of a royalty.



# **DRAFT KENYA STANDARD**



#### DKS 2501:2013 Foreword

This Kenya Standard was developed by the Technical Committee on Surface Active Agents under the guidance of the Standards Projects Committee and it is in accordance with the procedures of the Bureau.

This standard is intended for the evaluation of detergent-disinfectant liquid toilet cleansers based on quaternary ammonium compounds and other chemical agents that are miscible with water and intended for cleaning toilet bowls and urinals. Quaternary ammonium compounds are organically substituted ammonium compounds in which the nitrogen compound has a valence of five. Four of the substituent radicals are alkyl or heterocyclic radicals and the fifth is an ion. Examples include centrimide, Benzalkonium chloride and benzethonium chloride.

Using this standard, it is not possible to determine the bactericidal activity of the undiluted product. Some dilution is always produced by the addition of inoculum, hard water and sterile skimmed milk. If a product complies who the test requirements, it can be considered to be bactericidal.

During the preparation of this standard, reference was made to the following document: SANS 639:2009, Detergent-disinfectants based on quaternary ammonium compounds Acknowledgement is hereby made for assistance derived from this source.

## Detergent-disinfectant liquid toilet cleansers — Specification

#### 1 Scope

This Kenya Standard specifies requirements and test methods for detergent-disinfectant liquid toilet cleansers based on quaternary ammonium compounds and other chemical agents (such as compatible surface-active agents and phosphates) that are miscible with water and intended for cleaning toilet bowls and urinals.

#### 2 Normative references

This Kenya Standard incorporates by dated and undated reference, provisions from other publications. These normative references are cited at the appropriate place in the text and publications are listed hereafter. For dated references, subsequent amendments to or revision anv of these publications apply to this Kenya Standard only when incorporated in it by ame ent o revision. For undated references the latest edition of the publication referred to applies.

KS 929: Chemical disinfectants and antiseptics – Quantitative suspension test or the e evaluation of bactericidal activity of chemical disinfectants and antiseptics used in food, indus estic, and institutional areas - Test method and requirements (phase 2, step 1)

#### 2 Definitions

For the purposes of this standard the following definitions an

#### 2.1

#### detergent-disinfectant

Product that cleans and disinfects simultaneous

#### 2.2

5

#### disinfectant

chemical agent that kills most veget tive forms of pathogenic and other micro-organisms (but not necessarily all bacteria and fungal mycebacteria, or viruses) on inanimate surfaces pore

#### 3 Requirements

3.1 The detergent-disinfectant india toilet cleanser shall be a homogeneous liquid, free from sediment and suspended matter

3.2 ant liquid toilet cleanser shall have a stable, distinct colour. The detergen dish

The de 3.3 rgent-visinfectant liquid toilet cleanser shall be miscible in water in all proportions.

rgent-disinfectant liquid toilet cleanser shall be stable, and not deteriorate in storage 3.4 bt in its

rigina hed container at ambient temperature for a period of one year from the date of ınd anufacture.

The detergent-disinfectant liquid toilet cleanser shall not affect toilet bowl surfaces when used as directed by é manufacturer.

#### **Bactericidal efficacy** 3.6

When the detergent-disinfectant liquid toilet cleanser is tested in accordance with KS 929, at the prescribed

concentration by the manufacturer, it shall, within 5 minutes, kill at least 99.9 % of microorganisms.

3.7 The detergent-disinfectant liquid toilet cleanser shall comply with the requirements given in Table 1, when tested in accordance with the methods described therein.

#### Table 1: Requirements for detergent-disinfectant toilet cleansers

SI. No	Characteristic	Requirement	Test Method
i)	pH, max	10.5	
ii)	Surfactant, percent, m/m, min	0.2	Annex A
iii)	Bactericidal efficacy	to pass test	KS 929

### 4 Packaging and marking

### 4.1 Packing

Detergent-disinfectant liquid toilet cleansers shall be packed in plastic or other since op que containers that are strong enough to protect the product adequately during normal produing, transportation and storage.

#### 4.2 Marking

**4.2.1** The container shall be marked in prominent, legible and indefine markings with the following information:

**4.2.2** Individual containers shall be marked with the following information:

- i) manufacturer's name and physical address NOTE: The name, physical address of the distributor supplier and trade mark may be added as required
- ii) product name as "detergent-disinfectant wilet a
- iii) Ingredients
- iv) the net contents;
- v) the instructions for use;
- vi) the code number or bate number
- vii) date of manufacture and use by date
- viii) first aid instructions
- ix) country of or ain
- x) precautionary in tice marked in either English or Kiswahili languages with the following information:

he word CAUTION' shall be in a colour that contrasts with the surrounding for easy visuality and shall be followed with the statement, 'READ LABEL BEFORE USE';

e following shall be marked immediately under i):

- 'Keep Out of Reach of Children';
- the product shall be used for the cleaning of toilet bowls and urinals only;
- Do not mix with incompatible substances such as soap and detergents;

### 5 Sampling and compliance with the standard

5.1 Sampling

**5.1.1** For ascertaining the conformity of the lot to the requirements of this standard, tests shall be carried out on each lot separately.

**5.1.2** The number of packages and product units from each container respectively to be selected for drawing the sample shall be in accordance with table 2.

**5.1.3** The sample so drawn shall be deemed to represent the lot. From a given lot, for product units of 500  $\text{cm}^3$  or more a sample of one unit shall be chosen.

**5.1.4** Take at random the number of containers shown in column 2 of Table 2, relative to the appropriate lot size given in column 1.





### Annex A

#### (Normative)

#### Determination of non-ionic, anionic and cationic surfactant content

- A1. The molar mass and chemical name of all surfactants used shall be supplied by the manufacturer on request by Kenya Bureau of Standards, when use of high pressure liquid chromatography (HPLC) equipment is employed in the determination of surfactant content in the product.
- A2. Accurately weigh about 100 g of the compound in a 250-mL Erlenmeyer flask and neukalize with caustic soda solution to a pH of 8.8 to 7.0.
- A3. Evaporate the resulting solution to dryness at 105 °C. Cool to room temperature and wash the resulting solids with five 20 mL aliquots of chloroform, filtering and collection each in one tared 300-mL beaker. Evaporate the chloroform and determine the weight of organic solids,  $(W_1)$  g.

 $W_1$  (organic Solids) = <u>Mass of solids</u> Mass compound

- **NOTE:** You may use myer's reagent to confirm nonionic surfacent. Dissolve 100 mg to 150 mg of isolated surfactants in 5 mL of distilled water. Add 2 drops of myer's reagent. A yellow precipitate will form if non-usin surfactant is present.
- A4. Dissolve the organic solids in 50 mL of ethanovand pass resulting solution through a 22 mm x 200 mm x 250 mm chromatographic column of reshy regenerated cationic exchange resin (150 mm to 175 mm mesh). Wash the beaker with four 50 mL aliquots of ethanol collecting them in one tared beaker. (Cationic surfactance if present, will be retained, the eluate will contain nonionic and anionic surfactance it present). Evaporate the alcohol eluate in an oven. Weigh accurately, (*W*<sub>2</sub>).

Cationic surfactant  $g = V - V_{2}$ 

A5. Dissolve organic solids (16), 9 obtained in A4 in 50 mL of ethanol and pass the resulting solution through a chromat graphic column specified in A4 but having freshly regenerated anionic exchange resin (150 mm to 175 mm mesh). Wash the beaker with four 50 mL aliquots of alcohol collecting them in one tared beaker. (Anionic surfactant, if present, will be retained, the eluate will contain only nonionic surfactant). Weigh accurately,  $W_3$ , g.

A6. By difference,  
Anicy ic surjectant content, 
$$g = (W_2 - W_3)$$
  
Subjectant content per cent m/m  
 $= \frac{(W_1 - W_2) + W_3}{m} \times 100$  per cent (for cationic cleansers)  
OR  
 $= \frac{W_2}{m} \times 100$  per cent (anionic cleanser)

m = x is per cent (anionic cleanser)